

SYNTHESIS OF α -METHYLENE MONOSUBSTITUTED δ -LACTONES FROM α -PHOSPHONOLACTONES
NEW WITTIG-HORNER COMPOUNDS

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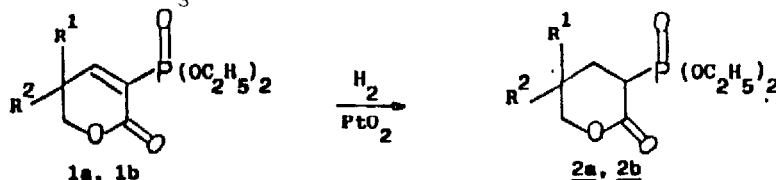
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Abstract- The synthesis of the α -methylene monosubstituted δ -lactones 6-10 and 14-18 from the α -phosphonolactones 2a, 2b and the aldehydes 3-5 and 11-13 is described.

The reaction of α -halolactones with trialkylphosphites¹⁻³ (Arbuzov reaction) represents a facile and elegant procedure for the synthesis of Wittig-Horner reagents. The general use of this reaction is limited by the small number of the α -halolactones obtainable^{4,5}. By hydrogenation of the α -phosphonolactones α, β -unsaturated 1a, 1b, obtained in our laboratory⁶, we have opened a new route to a class of compounds of synthetic interest.

The hydrogenation of the compounds 1a and 1b in the presence of platinum (IV) oxide in isobutyl alcohol (r.t) led to the unknown α -diethylphosphonolactones 2a ($R^1 = \text{CH}_3$, $R^2 = \text{C}_3\text{H}_7$) and 2b ($R^1 = R^2 = \text{CH}_3$) in 100% yields.



Scheme 1

The EI mass spectra of the α -phosphonolactones show the presence of molecular ions at $m/z = 292$ of 2a and $m/z = 264$ of 2b supporting the complete hydrogenation of the starting compounds 1a, 1b. Olefination of the 2a with the aldehydes 3-5 and of 2b with the aldehydes 11-13 under Wittig-Horner conditions gives the α -methylene monosubstituted δ -lactones 6-10 ($R^1 = \text{CH}_3, \text{C}_3\text{H}_7$) and 14-18 ($R^1 = R^2 = \text{CH}_3$) in 50-70% yields; ($E/Z=3/2$) (Scheme 2).

We have observed that the reaction of 2b with 4-nitrobenzaldehyde 11 and of 2a with the 5-methyl-thiophene-carbaldehyde 4 afforded only pure E -isomeres 8 and 14. In the other reactions E/Z -isomers have been always obtained and separated by flash chromatography.

In the $^1\text{H-NMR}$ spectra exocyclic protons are at a lower field - at $\delta = 7.98 - 8.13$ ppm for aromatic derivatives; $\delta = 6.70 - 7.10$ ppm for aliphatic derivatives - (s. Table 1) in the E -isomers by respect to Z -isomers - at $\delta = 6.23$ ppm for 6 and $\delta = 5.73 - 6.03$ ppm for aliphatic derivatives -. This difference in the chemical shifts of the olefinic protons in the E -

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